



Synthesis of 2-Substituted Indolo Pyridin-4-ones.

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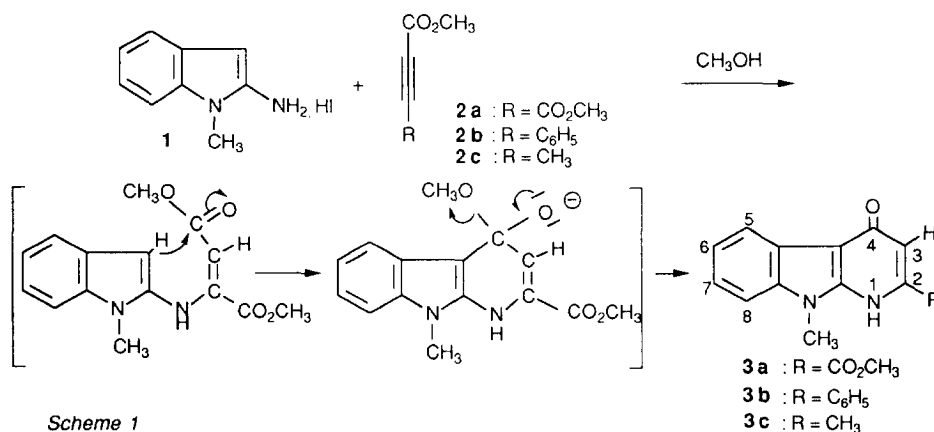
Summary. The synthesis of 2-substituted indolo-pyridin-4-ones was achieved in good yields by a one-pot reaction between *N*-Me amino-2-indole (as hydriodide) with acetylenic esters.

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In the course of a programme devoted to the search of bioactive compounds from marine invertebrates, two α -carboline, the cytotoxic grossularines-I and -II, isolated from the tunicate *Dendrodoa grossularia*¹ as the first naturally occurring α -carboline, are likely to act on DNA². This prompted us to undertake efforts towards the synthesis of these natural compounds and analogues. Several syntheses of α -carboline are described and recently a new approach to α -carboline bearing a dimethyl-amino-imidazole ring was proposed^{3,4}.

The synthesis of α -carboline in the grossularine series can be envisaged starting from indolo-pyridin-4-ones **3**. As previously reported, the condensation of *N*-methyl amino-2 indole (as hydriodide), with β -diketones⁵ or with diketenes yielded indolopyridin-2-ones accompanied, in this latter case, by minor amounts (16%) of indolopyridin-4-ones⁶, characterized on the basis of differences in the UV absorbance in respect to indolopyridin-2-ones. Hence another method to obtain indolo-pyridin-4-ones was desirable.

A general route to quinolone heterocycle synthesis was investigated, namely the Michael addition of a nucleophile moiety to an acetylenic carbonyl compound⁷. According to this scheme, condensation of anilines with acetylenic esters⁸, allenic esters⁹ or acetylenic derivatives devoid of carboxylate groups using Pd(0) catalyzed carbonylation¹⁰ led to useful methods for syntheses of 2-substituted quinolones. Hence we investigated the possibility of condensing *N*-substituted amino-2 indole with suitable acetylenic derivatives to obtain indolo-pyridinones. In this paper we describe the first unambiguous synthesis of indolo-pyridin-4-ones.



N-methyl amino-2 indole hydriodide **1** was chosen as the starting material, which was easily prepared according to¹¹. Condensation of **1** with dicarbomethoxy-2-acetylene **2a** was achieved by heating (reflux) during 5 hours a methanolic solution of **1** (1mM) with **2a** (1.2 mM). Workup furnished the indolo-pyridin-4-one **3a** in a 65% yield after purification on preparative t.l.c. (chloroform/acetone 7/3), (m.p. 264° C, M.S. M⁺ 256, IR: 3400, 1730, 1680 cm⁻¹). Direct (HMQC¹²) and long range (HMBC¹³) 2D ¹H-¹³C NMR

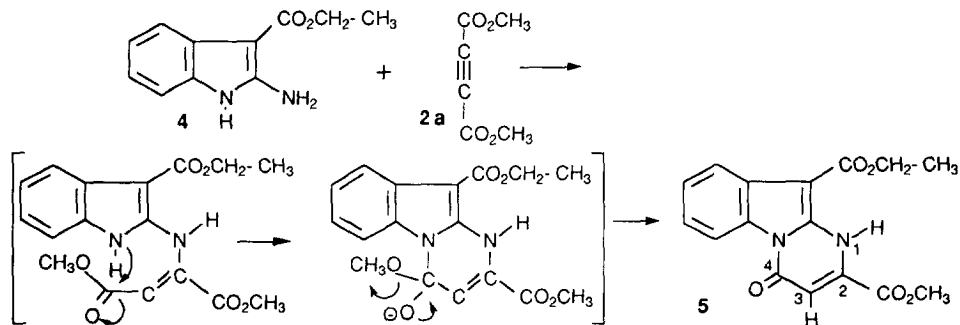
heteronuclear correlations confirmed the pyridin-4-one structure and led to assign all the proton and the carbon chemical shifts (Table). The reaction may proceed through an enamine adduct formed via a Michael addition, which undergoes cyclization to indolo-pyridinone **3a**. Other easily available acetylenic substrates were condensed with **1** by the same procedure to obtain the indolopyridinones : **3b** (R= C₆H₅)(65 %), m.p. 172°C, M.S. (C.I.) : [M+H]⁺ 275; **3c** (R= CH₃), 60%, m.p.178-180°C, M.S. (E.I.) M⁺ 212. (Scheme 1).

Table . ¹H (300 MHz) and ¹³C (75 MHz) NMR data of compounds **3a**, **3b**, **3c** (δ ppm)

Compounds	3a (DMSO d-6)(i)		3b (CDCl ₃)(ii)		3c (CDCl ₃)(iii)	
	¹ H	¹³ C	¹ H	¹³ C	¹ H	¹³ C
1a		151.4		150.9		154.2
2		134.7		147.0		147.5
3	6.91 (s)	102.0	6.40 (s)	100.3	6.98 (s)	109.3
4		162.3		160.2		160.2
4a		104.2		104.7		105.3
5a		119.2		119.2		123.7
5	8.45 (d, 1H, J = 8)	123.4	7.62 (dd)	124.1	7.52 (d)	126.8
6	7.24 (dd, 1H, J = 8,8)	119.4	7.62 (m, 1H)	120.3	7.33 (dd)	120.8
7	7.47 (dd, 1H, J = 8,8)	125.4	7.40	120.7	7.35 (dd)	122.4
8	7.58 (d, J=8)	109.3	7.0 (d)	108.9	7.05 (s)	111.7
8a		139.4		138.7		141.7
N-CH ₃	3.84 (s, 3H)	27.7	4.0	28.7	4.0	30.2

i) COOCH₃ δ 4.01 (52.7, 166.4. ii) C₆H₅ δ 7.52 (d, 2H), 7.35 (m, 3H), (126.4 (2C), 118.6 (2C), 128.7, 139.5. iii) CH₃ δ 2.4 (23.2).

When the starting material was a 3-substituted amino-2 indole such as the amino-2 carbethoxy-3 indole **4**, the condensation with **2a** afforded a pyrimido [3,2-a] indole-3-carboxylate **5**¹⁴ via the expected intramolecular cyclization occurring on the indolic NH (68%).



Scheme 2.

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14. ¹H NMR (CDCl₃, δ ppm): 8.62 (d) H-8; 7.78 (d) H-5; 7.42 (dd) H-7; 7.23 (dd) H-6; 6.52 (s) H-3; 4.42 (q) OCH₂-CH₃; 4.01 (s) COOME; 1.42 (t) OCH₂-CH₃. M.S. M⁺ 312.

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